

Fig. 3. $O R T E P$ stereo drawing of the molecular packing of the title compound in a unit cell.
reasonable ranges of magnitudes. The packing of the molecules in the crystal is shown in Fig. 3 and there are no intermolecular contacts less than van der Waals distances.

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## References

Chu, S. S. C. \& van der Helm, D. (1974). Acta Cryst. B30, 2489-2490.
Chu, S. S. C. \& van der Helm, D. (1975). Acta Cryst. B31, 1179-1183.
Chu, S. S. C. \& van der Helm, D. (1976). Acta Cryst. B32, 1012-1016.
Chu, S. S. C., Napoleone, V., Ternay, A. L. Jr \& Chang, S. (1982). Acta Cryst. B38, 2508-2511.

Chu, S. S. C. \& Yang, H. T. (1977). Acta Cryst. B33, 1892-1896.
Cremer, D. \& Pople, J. A. (1975). J. Am. Chem. Soc. 97, 1354-1358.
Cromer D. T. \& Liberman, D. (1970). J. Chem. Phys. 53, 1891-1896.
Germain, G., Main, P. \& Woolfson, M. M. (1971). Acta Cryst. A27, 368-376.
International Tables for X-ray Crystallography (1962). Vol. III, pp. 201-207. Birmingham: Kynoch Press.
Johnson, C. K. (1965). ORTEP. Report ORNL-3794. Oak Ridge National Laboratory, Tennessee.
Shiono, R. (1971). Tech. Rep. 49. Crystallography Department, Univ. of Pittsburgh.
Stewart, R. F., Davidson, E. R. \& Simpson, W. T. (1965). J. Chem. Phys. 42, 3175-3187.

# 1-Methylphenothiazine and 1-Ethylphenothiazine, a Nearly Isomorphous Pair 

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#### Abstract

Methylphenothiazine, $\mathrm{C}_{13} \mathrm{H}_{11} \mathrm{NS}$, monoclinic, $P 2_{1} / c, Z=4, M_{r}=213.30, a=9.063$ (1), $b=$ 8.968 (2), $\quad c=13.503$ (1) $\AA, \quad \beta=102.29(1)^{\circ}, \quad V=$ $1072.4(3) \AA^{3}, \quad D_{x}=1.321 \mathrm{Mg} \mathrm{m}^{-3}, \quad \mu(\mathrm{Cu} \mathrm{K} \mathrm{\alpha})=$ $2.28 \mathrm{~mm}^{-1}$. Final $R=0.053$ for 1658 observed reflections. 1-Ethylphenothiazine, $\mathrm{C}_{14} \mathrm{H}_{13} \mathrm{NS}$, monoclinic, $P 2_{1} / c, \quad Z=4, \quad M_{r}=227.33, \quad a=9.509(1), \quad b=$ 8.962 (1), $\quad c=14.036$ (2) $\AA, \quad \beta=105.34(1)^{\circ}, \quad V=$ 1153.6 (3) $\AA^{3}, \quad D_{x}=1.309 \mathrm{Mg} \mathrm{m}^{-3}, \quad \mu(\mathrm{Cu} K \alpha)=$ $2.15 \mathrm{~mm}^{-1}$. Final $R=0.035$ for 1731 observed reflections. The crystal structures of 1 -methylphenothiazine and l-ethylphenothiazine are nearly isomorphous. The folding angles are 154.8 and $147.4^{\circ}$ for 1 -methyl- and 1-ethylphenothiazines respectively.


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Introduction. These structures were determined as a continuous study of 1 - and 10 -substituted, and $1,10-$ disubstituted phenothiazines as described in the preceding paper (Chu, Napoleone, Ternay \& Chang, 1982). The preparation of 1 -methyl- and 1-ethylphenothiazines was carried out in the same manner as that of 1 -isopropylphenothiazine (Chu et al., 1982). Crude of 1 -methylphenothiazine was sublimed ( $383 \mathrm{~K}, 2.67 \mathrm{~Pa}$ ) to produce the desired material (m.p. $411-412 \mathrm{~K}$ ) in $42 \%$ yield. Crude of 1ethylphenothiazine was sublimed ( $443 \mathrm{~K}, 26.67 \mathrm{~Pa}$ ) with a $53 \%$ yield of the desired product (m.p. 394-396 K).

The unit-cell parameters of each compound were obtained from a least-squares analysis of 15 reflections with $2 \theta$ values measured on a Syntex $P 2_{1}$ automatic

Table 1. Experimental data

|  | -Methylphenothiazine | 1-Ethylphenothiazine |
| :---: | :---: | :---: |
| X-radiation used for data collection | $\lambda(\mathrm{CuK} \alpha)=1.5418 \AA$ | $\lambda(\mathrm{CuK})^{\prime}=1.5418 \AA$ |
| Total number of reflections $2 \theta_{\max }=130^{\circ}$ | ns 1778 | 1936 |
| Number of observed reflections $I>3 \sigma(I)$ | 1665 | 1737 |
| Number of reflections affected by extinction and omitted from the refinement | 7 | 6 |
| $2 \theta$ values used for the determination of unitcell parameters | $21-70^{\circ}$ | $64-86^{\circ}$ |
| Crystal size | $0.30 \times 0.58 \times 0.24 \mathrm{~mm}$ | $0.21 \times 0.45 \times 0.15 \mathrm{~mm}$ |
| Orientation of crystal along $\varphi$ axis of diffractometer | $b$ axis | $b$ axis |
| Disagreement index, $R$ | 0.053 | 0.035 |
| Weighted disagreement index $R_{w}$ | 0.059 | 0.036 |

diffractometer. The experimental data are given in Table 1. The intensity data were reduced to structure amplitudes by the application of Lorentz and polarization factors; no absorption corrections were applied.
The determination and refinement of the structures were carried out in the same way as those described in the preceding paper. The final $R$ index $\left(\sum\left|F_{o}\right|-\left|F_{c}\right| \mid\right.$ $\left.\sum\left|F_{o}\right|\right)$ was 0.053 , and the weighted disagreement index, $R_{w}$, was 0.059 for 1-methylphenothiazine. The $R$ and $R_{w}$ for 1-ethylphenothiazine are 0.035 and 0.036 respectively. The final atomic parameters are given in Table 2.*

Discussion. The identification of the atoms and the configuration of the molecules are shown in Fig. 1(a) and (b) for 1-methyl- and 1-ethylphenothiazines respectively. The two structures are nearly isomorphous as indicated by the unit-cell and atomic parameters. The molecules are folded with the central ring in a boat conformation. The torsion angles and the Cremer \& Pople (1975) puckering parameters are shown in Table 3. The folding angles between the least-squares planes of the two benzo rings are 154.8 and $147.4^{\circ}$ for 1 -methyl- and 1-ethylphenothiazines respectively. The folding angle is $146 \cdot 1^{\circ}$ for 1 -isopropylphenothiazine as shown in the preceding paper. Similarly to 10 alkylphenothiazines (Chu \& van der Helm, 1975), the magnitude of the folding angle decreases with the increasing size of the alkyl substituent.

[^1]Table 2. Fractional atomic coordinates $\left(\times 10^{4}, \times 10^{3}\right.$ for H ) and isotropic thermal parameters
E.s.d.'s are given in parentheses and refer to the last positions of respective values.
$B_{\text {eq }}$ is calculated from the relation: $B_{\text {eq }}=\frac{1}{3} \sum \sum B_{l j} a_{i}{ }^{*} a_{j}{ }^{*} \mathbf{a}_{l} \cdot \mathbf{a}_{j}$.

| 1-Methylphenothiazine |  | $y$ | $z$ | $B_{\text {eq }} / B\left(\AA^{2}\right)$ |
| :---: | :---: | :---: | :---: | :---: |
|  |  |  |  |
| S(5) | 6911 (1) |  | 1769 (1) | 1342 (1) | 4.59 (3) |
| $N(10)$ | 5041 (2) | 4177 (3) | 2027 (2) | 3.93 (9) |
| C(1) | 3119 (3) | 4224 (3) | 466 (2) | 3.93 (11) |
| C(2) | 2590 (3) | 3680 (3) | -508 (2) | 4.43 (12) |
| C(3) | 3395 (3) | 2629 (3) | -932 (2) | 4.80 (13) |
| C(4) | 4764 (3) | 2119 (3) | -378(2) | 4.45 (13) |
| C(6) | 9120 (3) | 3300 (3) | 2646 (2) | 4.53 (12) |
| C(7) | 9652 (3) | 4337 (3) | 3394 (2) | $5 \cdot 05$ (14) |
| C(8) | 8654 (3) | 5334 (3) | 3673 (2) | 4.81 (13) |
| C(9) | 7133 (3) | 5301 (3) | 3207 (2) | 4.24 (12) |
| C(11) | 4493 (3) | 3698 (3) | 1028 (2) | 3.44 (10) |
| C(12) | 5288 (3) | 2624 (3) | 608 (2) | 3.72 (11) |
| C(13) | 7594 (3) | 3225 (3) | 2199 (2) | 3.72 (11) |
| C(14) | 6588 (3) | 4240 (3) | 2469 (2) | 3.47 (10) |
| C(15) | 2215 (3) | 5374 (4) | 904 (2) | $5 \cdot 34$ (15) |
| $\mathrm{H}(\mathrm{N})$ | 452 (3) | 497 (4) | 230 (2) | 6.9 (8) |
| H(2) | 158 (3) | 401 (3) | -90 (2) | 5.4 (7) |
| H(3) | 306 (3) | 222 (4) | -170 (2) | $7 \cdot 5$ (9) |
| H(4) | 539 (3) | 144 (3) | -66 (2) | 4.9 (7) |
| H(6) | 980 (3) | 247 (4) | 244 (2) | $6 \cdot 3$ (7) |
| H(7) | 1083 (3) | 441 (3) | 379 (2) | $5 \cdot 8$ (7) |
| H(8) | 898 (3) | 611 (4) | 428 (2) | 7.4 (9) |
| H(9) | 642 (3) | 594 (3) | 339 (2) | $4 \cdot 2$ (6) |
| H(15) 1 | 286 (4) | 629 (4) | 124 (3) | $9 \cdot 1$ (10) |
| H(15)2 | 125 (4) | 565 (4) | 38 (3) | 8.6 (10) |
| H(15)3 | 186 (3) | 494 (4) | 155 (2) | $6 \cdot 7$ (8) |


| 1-Ethylphenothiazine (S-atom coordinates $\times 10^{5}$ ) |  |  |  |  |
| :---: | :---: | :---: | :---: | :---: |
| S(5) | 69090 (5) | 15598 (5) | 14670 (4) | 4.40 (1) |
| $\mathrm{N}(10)$ | 5131 (1) | 3974 (2) | 2029 (1) | 3.82 (6) |
| C(1) | 3390 (2) | 4179 (2) | 401 (1) | 3.67 (7) |
| C(2) | 2951 (2) | 3695 (2) | -574 (1) | 4.38 (8) |
| C(3) | 3732 (2) | 2634 (2) | -933 (1) | 4.78 (9) |
| C(4) | 4966 (2) | 2029 (2) | -322 (1) | 4.49 (8) |
| C(6) | 9078 (2) | 3157 (2) | 2733 (1) | 4.64 (9) |
| C(7) | 9593 (2) | 4221 (2) | 3448 (2) | 5.35 (10) |
| C(8) | 8627 (2) | 5214 (2) | 3693 (1) | $5 \cdot 14$ (9) |
| C(9) | 7159 (2) | 5157 (2) | 3214 (1) | 4.42 (8) |
| C(11) | 4639 (2) | 3555 (2) | 1027 (1) | 3.37 (6) |
| C(12) | 5411 (2) | 2464 (2) | 663 (1) | 3.63 (7) |
| C(13) | 7597 (2) | 3053 (2) | 2281 (1) | 3.77 (7) |
| C(14) | 6618 (2) | 4070 (2) | 2501 (1) | 3.55 (7) |
| C(15) | 2464 (2) | 5307 (2) | 759 (1) | 4.30 (8) |
| C(16) | 1375 (2) | 4537 (2) | 1225 (2) | 5.39 (10) |
| $\mathrm{H}(\mathrm{N})$ | 460 (2) | 467 (2) | 220 (1) | $4 \cdot 7$ (4) |
| H(2) | 208 (2) | 418 (2) | -100 (1) | $4 \cdot 3$ (4) |
| H(3) | 348 (2) | 236 (2) | -165 (1) | 6.5 (5) |
| H(4) | 558 (2) | 129 (2) | -55 (1) | $5 \cdot 2$ (4) |
| H(6) | 979 (2) | 243 (2) | 258 (1) | $5 \cdot 2$ (4) |
| H(7) | 1062 (2) | 427 (2) | 383 (1) | $7 \cdot 1$ (6) |
| H(8) | 897 (2) | 601 (2) | 424 (1) | $6 \cdot 8$ (5) |
| H(9) | 647 (2) | 587 (2) | 335 (1) | 5.4 (5) |
| H(15) 1 | 183 (2) | 600 (2) | 16 (1) | $5 \cdot 0$ (4) |
| H(15)2 | 311 (2) | 600 (2) | 125 (1) | 4.9 (4) |
| H(16)1 | 67 (2) | 380 (2) | 70 (1) | 7.4 (6) |
| $\mathrm{H}(16) 2$ | 190 (2) | 384 (2) | 183 (1) | 7.4 (6) |
| H(16)3 | 71 (2) | 533 (3) | 140 (2) | 8.4 (6) |

The $\mathrm{C}(15)$ of the methyl substituent in 1 methylphenothiazine is essentially coplanar with the benzo ring and the deviation of $\mathrm{C}(15)$ from the least-squares plane of the benzo ring is $0.008 \AA$. However, the deviation of $\mathrm{C}(15)$ from the least-squares

(a)

(b)

Fig. 1. ORTEP drawings (Johnson, 1965) of one molecule of (a) 1-methylphenothiazine and (b) 1-ethylphenothiazine.


Fig. 2. Bond length ( $\AA$ ) and bond angles ( ${ }^{\circ}$ ) of (a) 1methylphenothiazine and (b) 1 -ethylphenothiazine. The e.s.d.'s are in parentheses.

Table 3. Conformation of the central ring
Cremer \& Pople (1975) puckering parameters

|  |  |  |  |
| :---: | :---: | :---: | :---: |
| $q_{2}$ | Ideal boat | l-Methylphenothiazine | l-Ethylphenothiazine |
| $q_{3}$ | $Q \sin \theta$ | $0.450 \AA$ | $0.523 \AA$ |
| $Q$ |  | 0.069 | 0.078 |
| $\varphi$ | 0 or $180^{\circ}$ | 0.455 | 0.529 |
| $\theta$ | 90 or $270^{\circ}$ | $179.9^{\circ}$ | $181.1^{\circ}$ |
|  |  | 81.3 | 81.5 |

Torsion angles
1-Methylphenothiazine 1-Ethylphenothiazine

| $\mathrm{S}(5)-\mathrm{C}(13)-\mathrm{C}(14)-\mathrm{N}(10)$ | $-3.6(4)^{\circ}$ | $-4.8(2)^{\circ}$ |
| :--- | ---: | ---: |
| $\mathrm{C}(13)-\mathrm{C}(14)-\mathrm{N}(10)-\mathrm{C}(11)$ | $-32.2(4)$ | $-36.8(2)$ |
| $\mathrm{C}(14)-\mathrm{N}(10)-\mathrm{C}(11)-\mathrm{C}(12)$ | $31.8(4)$ | $37.3(2)$ |
| $\mathrm{N}(10)-\mathrm{C}(11)-\mathrm{C}(12)-\mathrm{S}(5)$ | $4.1(4)$ | $3.6(2)$ |
| $\mathrm{C}(11)-\mathrm{C}(12)-\mathrm{S}(5)-\mathrm{C}(13)$ | $-30.1(3)$ | $-34.1(2)$ |
| $\mathrm{C}(12)-\mathrm{S}(5)-\mathrm{C}(13)-\mathrm{C}(14)$ | $29.6(2)$ | $34.6(2)$ |



Fig. 3. ORTEP stereodrawing of the molecular packing of 1ethylphenothiazine in a unit cell.
plane of the benzo ring is $0.048 \AA$ in 1 ethylphenothiazine, and is $0.098 \AA$ in 1 isopropylphenothiazine (Chu et al., 1982). The nonplanarity of the alkyl substituent is due to the non-bonded interaction when the size of the alkyl substituent increases. The conformation of the ethyl substituent in 1-ethylphenothiazine is similar to that of the $C(15)-C(16)$ bond in 1 -isopropylphenothiazine. The torsion angles of $\mathrm{C}(2)-\mathrm{C}(1)-\mathrm{C}(15)-\mathrm{C}(16)$ are $93 \cdot 1$ (2) (it is $-93 \cdot 0^{\circ}$ for the mirror-related molecule since 1-ethylphenothiazine belongs to a centrosymmetric space group) and $-94.2(5)^{\circ}$ in 1 ethylphenothiazine and 1 -isopropylphenothiazine respectively.

The bond lengths and bond angles with their standard deviations for 1-methyl- and 1ethylphenothiazines are shown in Fig. 2(a) and (b) respectively. The $\mathrm{C}-\mathrm{S}$ and $\mathrm{C}-\mathrm{N}$ bond lengths within the central ring are in good agreement with those observed in other phenothiazines (Chu \& van der Helm, 1975, 1976). The size of the $\mathrm{C}-\mathrm{S}-\mathrm{C}$ and $\mathrm{C}-\mathrm{N}-\mathrm{C}$ bond angles within the central ring varies with
the size of the 1 -alkyl substituents and is correlated with the size of the folding angle as in other phenothiazines. All the $\mathrm{C}-\mathrm{H}$ bond lengths and $\mathrm{C}-\mathrm{C}-\mathrm{H}$ and $\mathrm{H}-\mathrm{C}-\mathrm{H}$ bond angles are within reasonable ranges of magnitudes. The packing of 1 -ethylphenothiazine in the crystal is shown in Fig. 3. There are no intermolecular contacts less than van der Waals distances.

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## References

Chu S. S. C. \& van der Helm, D. (1975). Acta Cryst. B31, 1179-1183.

Chu, S. S. C. \& van der Helm, D. (1976). Acta Cryst. B32, 1012-1016.
Chu, S. S. C., Napoleone, V., Ternay, A. L. Jr \& Chang, S. (1982). Acta Cryt. B38, 2506-2508.

Cremer, D. \& Pople, J. A. (1975). J. Am. Chem. Soc. 97, 1354-1358.
Johnson, C. K. (1965). ORTEP. Report ORNL-3794. Oak Ridge National Laboratory, Tennessee.

Acta Cryst. (1982). B38, 2511-2513

# Methyl 4,6-Bis( $O$-p-chlorobenzoyl)-2,3-dideoxy-3-C-(methoxycarbonylmethyl)- $\alpha$-D-ribohexopyranoside 

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#### Abstract

Crystals of the title compound, $\mathrm{C}_{24} \mathrm{H}_{24} \mathrm{Cl}_{2} \mathrm{O}_{8}$, $M_{r}=511.35$, are monoclinic, $P 2_{1}, a=5.752$ (3), $b=$ 15.436 (3), $c=13.698$ (3) $\AA, \beta=93.74$ (3) ${ }^{\circ}, D_{m}=$ $1.43, D_{x}=1.40 \mathrm{Mg} \mathrm{m}^{-3}, Z=2$. The structure was determined from $\mathrm{Cu} K \alpha$ diffractometer data (with great difficulty), and refined to $R=0.042$ for 898 reflexions. The six-membered pyranose ring is in the chair conformation, with the bulkier $4,6-p$-chlorobenzoyl substituents equatorial, and the $1-\mathrm{OMe}$ and $3-\mathrm{CH}_{2} \mathrm{CO}_{2} \mathrm{Me}$ groups axial. Bond lengths and angles and intermolecular distances are normal. The extremities of the $p$-chlorobenzoyl groups exhibit large thermal librations [r.m.s. displacement as large as 0.56 (1) $\AA$ for one Cl].


Introduction. The title compound [(1), $R=$ $\mathrm{ClC}_{6} \mathrm{H}_{4} \mathrm{CO}-J$ is a branched-chain sugar which was of interest because of possible antibiotic properties, and an X-ray crystal analysis was undertaken to verify structural assignments based on NMR data (Rosenthal \& Catsoulacos, 1968). The X-ray study was begun in

[^2]1967 (by DLH shortly after arrival in Vancouver from Antarctica, after a journey by Land Rover through South America).


The unit-cell dimensions were obtained by leastsquares refinement based on $2 \theta$ values for 30 reflexions, and intensities were measured on a Datex automated GE XRD-6 diffractometer with nickel-filtered $\mathrm{Cu} K a$ radiation and a $\theta-2 \theta$ scan technique. The scan speed was $1^{\circ} \min ^{-1}$, with 40 s background measurements before and after the scan. The intensities exhibited a rapid decrease with increasing $\theta$, so that the data set was rather limited. Of 1028 reflexions in the range $0<$ $\theta \leq 45^{\circ}, 898$ had $I / \sigma(I)>2 \cdot 0$, where $\sigma(I)=S+B+$ $(0.02 \mathrm{~S})^{2}, S=$ scan and $B=$ normalized background count. Crystal dimensions were $0.25 \times 0.05 \times 0.04$ mm , and no absorption corrections were considered necessary ( $\mu=28 \mathrm{~cm}^{-1}$ ).

Early attempts to determine the structure involved Patterson methods and efforts to pack the $p$-chloro-


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[^1]:    * Lists of structure factors and anisotropic thermal parameters have been deposited with the British Library Lending Division as Supplementary Publication No. SUP 36852 ( 24 pp.). Copies may be obtained through The Executive Secretary, International Union of Crystallography, 5 Abbey Square, Chester CH 1 2HU, England.

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